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Substituting  
and multiplying by the  
factor 10<sup>9</sup> for the  
density unit kg/km<sup>3</sup>,

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the mass of the atmosphere is determined to be  $m = 5.092 \times 10^{18}$  kg

Discussion

Performing the analysis with excel would yield exactly the same results. EES Solution for final result

$a = 1.2025166$   
 $b = -0.10167$   
 $c = 0.0022375$   
 $r = 6377$   
 $h = 25$   
 $m = 4 * \pi * (a * r^2 * h + r * (2 * a + b * r) * h^2 / 2 +$

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$a+2*b*r+c*r^2)*h^3/3+$   
 $(b+2*c*r)*h^4/4+c*h^5$   
 $/5)*1E+9$  1-7 Pressure,  
Manometer, and  
Barometer 1-34 C The  
pressure relative to  
the atmospheric  
pressure is called ...

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thermodynamics is  
based on

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observations whereas  
statistical

thermodynamics is.

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The effect of the initial pressure of

refrigerant-134 on the

volume of the tank is

to be investigated as

the initial pressure

varies from 0.5 MPa

to 1.5 MPa. The

volume of the tank is

to be plotted versus

the initial pressure,

and the results are to

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be discussed.

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If all phases can be  
represented on a one-

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dimensional diagram  
(that is, a straight line  
representing  
composition), we are  
dealing with a two  
component system  
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The results are:  $\rho(z) =$   
 $a + bz + cz^2 =$   
 $1.20252 + 0.101674z$   
 $+ 0.0022375z^2$  for  
the unit of  $\text{kg/m}^3$  ,  
(or,  $\rho(z) = (1.20252 +$   
 $0.101674z +$   
 $0.0022375z^2) \times 10^9$

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for the unit of  $\text{kg}/\text{m}^3$   
) where  $z$  is the  
vertical distance from  
the earth surface at  
sea level. At  $z = 7 \text{ km}$ ,  
the equation would  
give  $\rho = 0.60 \text{ kg}/\text{m}^3$  .

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